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apply the

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Framework (of purp

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ose/task/criteria) in

contexts including

assignments,

curricula,

assessment and

strategic initiatives,

all toward the goal

of enhancing

student success

equitably.

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Examples and  
Resources

a. potassium-39 b.  
potassium-40 c.  
potassium-41 7.

Write an equation-  
showing-the-  
relationship

between an atom's  
atomic number and  
its mass number.

8. Lithium has two  
isotopes: lithium-6

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and lithium-7. Draw a diagram, like those shown on the transparency, for each lithium isotope. Label the protons, electrons, neutrons, and

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of an s orbital?

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5.2 2. What is the  
relationship  
between the size of  
an s orbital and the  
principal energy  
level in which it is  
found? 3. What is

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the shape of a p orbital? How many p orbitals are there in a sublevel? 4.

Answers

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Masters Protons



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Neutrons Electrons

Potassium-39 19

20 19

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21 19

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22 19 19e 2 19p 1

19p 1 19p 1 19e 2

19e 2 20n 0 21n 0

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40 50 60 70 80 90

100 Solubility (g  
solute/100 g H<sub>2</sub>O)

90 100 80 70 60 50

40 30 20 10 0

Temperature (oC)

Solubilities as a

Function of

Temperature NaCl

KClO<sub>3</sub> KCl CaCl<sub>2</sub>

Ce<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> Solubil

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Formation of Ions

1. calcium and  
oxygen 2. Yes;

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each contains  
equal numbers of  
protons and  
electrons. 3.

ionization energy

4. positive; cation

5. negative; anion

6. No; it is not a

stable octet of

electrons. 7. No; it

is not a stable

octet of electrons.

8.

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